Nitrogen oxide gas emissions from temperate forest soils receiving long-term nitrogen inputs

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Abstract

From spring 2000 through fall 2001, we measured nitric oxide (NO) and nitrous oxide (N₂O) fluxes in two temperate forest sites in Massachusetts, USA that have been treated since 1988 with different levels of nitrogen (N) to simulate elevated rates of atmospheric N deposition. Plots within a pine stand that were treated with either 50 or 150 kg N ha⁻¹ yr⁻¹ above background displayed consistently elevated NO fluxes (100–200 μg N m⁻² h⁻¹) compared to control plots, while only the higher N treatment plot within a mixed hardwood stand displayed similarly elevated NO fluxes. Annual NO emissions estimated from monthly sampling accounted for 3.0–3.7% of N inputs to the high-N plots and 8.3% of inputs to the Pine low-N plot. Nitrous oxide fluxes in the N-treated plots were generally < 10% of NO fluxes. Net nitrification rates (NRs) and NO production rates measured in the laboratory displayed patterns that were consistent with field NO fluxes. Total N oxide gas flux was positively correlated with contemporaneous measurements of NR and NO₃⁻ concentration. Acetylene inhibited both nitrification and NO production, indicating that autotrophic nitrification was responsible for the elevated NO production. Soil pH was negatively correlated with N deposition rate. Low levels (3–11 μg N kg⁻¹) of nitrite (NO₂⁻) were detected in mineral soils from both sites. Kinetic models describing NO production as a function of the protonated form of NO₂⁻ (nitrous acid [HNO₂]) adequately described the mineral soil data. The results indicate that atmospheric deposition may generate losses of gaseous NO from forest soils by promoting nitrification, and that the response may vary significantly between forest types under similar climatic regimes. The lowering of pH resulting from nitrification and/or directly from deposition may also play a role by promoting reactions involving HNO₂.

Keywords: chemodetrification, N deposition, nitric oxide, nitrification, nitrous oxide

Received 22 July 2002; revised version received 1 October 2002 and accepted 4 November 2002

Introduction

Many forests in Europe and North America continue to receive elevated levels of atmospheric N deposition, deriving mainly from fossil fuel combustion and fertilizer production and use (Fenn et al., 1998; Gundersen et al., 1998). Total N deposition rates can range from above 50 kg N ha⁻¹ yr⁻¹ in high elevation sites downwind of industrial or agricultural areas to below 3 kg N ha⁻¹ yr⁻¹ in remote forests (NADP, 2002; Lovett et al., 1982; Tietema, 1993). Significant increases in N deposition rates worldwide have been predicted based on projected increases in energy and fertilizer consumption (Galloway et al., 1994; Hall & Matson, 1999). There is increasing concern that anthropogenic N inputs may not only exceed plant uptake capacity but may also have deleterious impacts on entire ecosystems. Potential effects include soil acidification (van Breemen et al., 1982), nutrient imbalances and losses (Johnson et al., 1991), nitrate (NO₃⁻) leaching to groundwater and streams (Kahl et al., 1993), soil emissions of N oxide gases.

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(Skiba et al., 1999) and, ultimately, forest decline (Aber et al., 1998). More information is needed regarding the partitioning of N inputs between different soil and plant pools, the variation in responses among forest types, and the effect on soil N cycling processes such as nitrification, denitrification, and N trace gas production.

While soil-to-atmosphere emissions of N oxide gases from forest soils have been proposed as a likely response to persistent N additions (Aber et al., 1998; Fenn et al., 1998), there have been only a few measurements of N₂O flux (Brumme & Beese, 1992; Klemetsson et al., 1997; Peterjohn et al., 1998), and fewer measurements of both N₂O and NO (Butterbach-Bahl et al., 1997; Skiba et al., 1999), in response to N deposition in temperate forests. Quantification of NO and N₂O emissions may help to improve total ecosystem N budget estimates for N-impacted forests (Magill et al., 2000). Soil emissions of N oxide gases themselves may have important impacts, including possible effects on regional tropospheric ozone levels, contribution to downwind N deposition, enhanced destruction of stratospheric ozone, and increases in total greenhouse gases (Crutzen, 1979, 1981).

The chronic-nitrogen-addition experiment at the Harvard Forest (HF) in central Massachusetts, USA is one of the longest running and most intensive studies of N deposition in temperate forest ecosystems. Since 1988, a variety of plant, soil, and whole ecosystem responses to experimental N additions have been monitored in two adjacent sites comprised of mixed hardwood and red pine, respectively (Bowden et al., 1991; Aber et al., 1993; Magill et al., 1997, 2000; Nadelhoffer et al., 1999). While small increases in N₂O emissions have been observed in the pine forest, N₂O emissions have been estimated to account for <0.4% of the total ecosystem N budget at HF (Magill et al., 1997, 2000). Increases in soil nitrification rates in response to N inputs have also been observed at HF, with the pine forest responding sooner than the hardwood forest (Magill et al., 2000). These results suggest that nitrification-driven emissions of NO (and possibly N₂O) may also be increasing, since previous studies in forest soils have found strong correlations between nitrification potential and N oxide fluxes (Verchot et al., 1999; Davidson et al., 2000). However, emissions of NO had not been measured at HF prior to the present study.

The objectives of the present study were (i) to make updated field measurements of N₂O flux and first-time measurements of NO flux from long-term N deposition study plots at HF, (ii) to compare fluxes between the mixed hardwood and red pine plots, (iii) to examine the influence of climatic factors on emissions rates, and (iv) to investigate the microbial and chemical processes responsible for the elevated NO emissions which were observed.

Methods and materials

Site description and long-term experimental design

Experiments were conducted at the chronic-N addition plots at HF in central Massachusetts, USA (42°30’N, 72°10’W), which is part of the National Science Foundation Long-Term Ecological Research network. Since 1988, two adjacent forested areas have received additions of NH₄NO₃ to simulate atmospheric N deposition in excess of background wet plus dry deposition of 8 kg N ha⁻¹ yr⁻¹ (Magill et al., 2000). One of the sites (hereafter referred to as ‘Hardwoods’) is comprised of a mixture of deciduous tree species that regenerated following clear cutting ca. 1950. The forest is dominated by black oak (Quercus velutina Lam.) and red oak (Q. rubra L.). The other forest (hereafter referred to as ‘Pines’) is comprised primarily of even-aged red pine (Pinus resinosa Ait.) planted in 1926. Experimental areas within each forest are divided into four 30 m × 30 m (0.09 ha) plots, and each plot is further subdivided into 5 m × 5 m (0.0025 ha) subplots. Annual treatments are as follows: (i) control plot (no additions), (ii) low-N addition plot (+50 kg N ha⁻¹ yr⁻¹), and (iii) high-N addition plot (+150 kg N ha⁻¹ yr⁻¹). Applications of NH₄NO₃ solution are made in six equal treatments beginning the first week of May and then at 4-week intervals ending in mid-September. Further experimental details are described by Magill et al. (1997). The low-N treatment level was selected to represent the upper range of actual N deposition rates (Lovett et al., 1982; Tietema, 1993; Gundersen et al., 1998). The high-N treatment level was employed in order to explore hypotheses related to the concept of N saturation (Aber et al., 1989, 1998; Fenn et al., 1998). The objective was to accelerate ecosystem responses that otherwise might occur over several decades, so they might become evident within a more practical experimental period (10–20 years).

Field NO and N₂O fluxes

Within each plot, three interior subplots were randomly selected for closed chamber NO and N₂O flux measurements. Plot locations were fixed for the entire study. Field gas fluxes were measured at approximately monthly intervals during snow-free periods beginning 11 June 2000 and concluding 24 November 2001. During each NO measurement, soil temperatures at 10 and 50 mm below the soil surface were measured using soil temperature probes (Fisher Scientific). Monthly flux measurements were made during mid-morning to early afternoon (10:00–14:00 local time), and were timed to occur when soil temperatures in the upper 50 mm were approximately equal to the mean of minimum and
maximum daily values. All reported monthly gas flux data were collected at least 7 days after N application, except the August 2000 and September 2000 data, which were collected 2 days after N application.

The gas flux chamber design was identical to that used in previous studies of N2O fluxes at HF (Bowden et al., 1990, 1991). Chambers consisted of 287-mm diameter (ID) by 40 mm high polyvinyl chloride (PVC) cylinders, which were placed on permanently installed PVC base rings immediately prior to measurement. At sampling intervals of approximately 15, 30 and 60 min following placement of the chamber, 9-mL gas samples were collected using polypropylene syringes from gas sampling ports in the centre of the chamber top. Replicate samples of ambient air taken during measurement periods were used as time 0 samples. Samples were transferred to evacuated glass vials, which were stored at room temperature prior to N2O analysis by gas chromatography (GC) with electron capture detection. Flux of N2O was calculated from the linear rate of change in N2O concentration, the chamber internal volume and soil surface area.

For NO flux measurement, a chamber of the same dimensions as above fitted with inlet and outlet fittings were used. Upon chamber placement, a continuous gas stream (0.03–0.09 m³ h⁻¹) was delivered to the chamber and delivered to a chemiluminescent N2O (NO + NO₂) analyzer (Unisearch Models LMA-3 and LMA-3D) using a vacuum pump within the analyzer. Prior to entering the analyzer, chamber gas was passed through granular CrO₃, which converts NO to NO₂. Outlet air from the analyzer was passed through KMnO₄-impregnated alumina granules, which act as a NO₂ scrubber (Purafil, Inc.) and anhydrous CaSO₄ (desiccant) prior to return to the chamber. In this configuration, the analyzer detects total NOx. Measurements were periodically made with bypass of the CrO₃ converter to confirm that emissions of NOx consisted mainly (> 99%) of NO and not NO₂. Therefore, we hereafter use the term ‘NO flux’ to denote flux measured as total NOx. Concentrations of NOx in the recirculating gas were recorded at 10–30 s intervals for 4–5 min after placement of the chamber top. The analyzer was calibrated within 1–24 h prior to measurements using gas streams containing NO in the range of 1–100 ppbv. Fluxes of NO (F_NOx, μg N m⁻² h⁻¹) were calculated from

$$F_{NO} = \frac{dC}{dt} \frac{V}{A} + C_A \frac{Q}{A}$$  \hspace{1cm} (1)

where dC/dt is the rate of change of NO concentration (μg N m⁻²) over time (h) determined by linear regression, V is the internal chamber volume (m³), A is the soil surface area (0.065 m²), C_A is the average NO concentration during the time interval of regression, and Q is the recirculation air flow rate (m³ h⁻¹). Deriving from mass balance considerations, the final term in Eqn 1 accounts for the removal of NO from the chamber headspace during recirculation.

Additional experiments were done to examine responses in gas fluxes to experimental N addition or rain events, and to record daytime fluctuations in gas fluxes on an hourly time-scale. A previous study at this site found no short-term response in N₂O emission to NH₄NO₃ applications over a period of 1–24 days (Magill et al., 1997). In the present study, field NO fluxes were measured 48 h and 1 h prior to, and 24 h, 48 h and 72 h following the NH₄NO₃ application of 29 May 2001. The response of NO fluxes to wetting was examined in August and October 2001. After measuring baseline gas fluxes, a volume of deionized water equivalent to a 25-mm rainfall event was applied to the area inside of and surrounding each chamber base ring to a distance of 0.25 m with a manual sprinkler. In August 2001, gas fluxes were measured 6 and 30 h after water was applied. In October 2001, fluxes were measured 24 h after water was applied and gas fluxes were measured only in plots which had previously displayed elevated NO fluxes. No actual rainfall was recorded during the 48 h preceding each wetting experiment. Daytime fluctuations in NO flux and soil temperatures were examined on two days in 2001 (3 July and 27 October). On these dates, measurements were made in the morning (7:00–8:30 LT), mid-day (12:00–13:30 LT) and afternoon (16:00–18:00 LT). The experiments examining daytime fluctuations and responses to N addition and wetting focused on NO emissions. Previous studies at HF have closely examined the dynamics of N₂O flux (Bowden et al., 1990, 1991; Magill et al., 1997).

**Soil sampling and analysis**

On the same day of each monthly gas flux sampling, soil samples (5–10 g each) were taken by hand trowel from 3 to 4 interior subplots and mixed together to generate two composite samples, one each from the organic (Oₐ + Oₜ) and mineral horizons. These samples were dried in the laboratory at 105 °C (mineral soil) or 65 °C (organic soil) for 24–48 h in order to determine gravimetric soil water content. Additional soil samples were collected in summer 2000 (24 August) and spring 2001 (1 June) for determination of inorganic N concentrations, soil pH and rates of net nitrification, N mineralization and NO production. These samples were taken at least 72 h following the August 2000 and May 2001 N applications from interior subplots other than those used for gas flux sampling. Prior to sampling, the upper layer (approximately 50–200 mm thick) of litter (Oₜ horizon) was removed from the sampling area. A section of split PVC pipe (50-mm ID × 200-mm long) was then pounded into the soil to a

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depth of 150 mm. Each core sample was separated into organic (O<sub>e</sub> + O<sub>h</sub> horizon) and mineral material, and delivered to the laboratory for processing within 1–3 days of collection.

Samples were first passed through a 6-mm-mesh sieve. Two subsamples (~5 g each) of mineral and organic sample were extracted for 1–2 h in 20 mL of 2 M KCl. Extracts were filtered through Whatman No. 42 paper and stored at 4°C prior to determination of ammonium-N (NH<sub>4</sub><sup>+</sup> – N) and nitrate plus nitrite-N (NO<sub>3</sub>– N + NO<sub>2</sub>– N) using an automated colorimetric analyzer (Perstorp Analytical, 3000 series). On the same day, two additional subsamples (10–30 g) were placed into separate 250-mL glass jars, sealed loosely, and incubated in a humid atmosphere at 20°C. After 14 days, a subsample (~5 g) was extracted with 2 M KCl as above. Potential net nitrification rates were calculated from the net increase in NO<sub>3</sub>– N – N + NO<sub>2</sub>– N concentrations, and potential net N mineralization rates were calculated from the net increase in total inorganic N occurring during the incubation period. Rates were expressed on a dry soil mass basis (μg N kg<sup>-1</sup> h<sup>-1</sup>). Soil pH was measured in separate subsamples of soils collected in August 2000 by manually mixing soil with 1 M KCl at a soil-to-liquid mass ratio of 2:1 (mineral soil) or 5:1 (organic soil). After settling for 30–90 min, solution was poured off for determination of pH using a combination electrode.

Instantaneous rates of NO production in the incubating soils were determined after 3–4 days of incubation by sealing each jar with a specially fitted lid attached to a dynamic flow-through system, which allowed for the continuous delivery of a humidified air stream through the jar prior to entering the NO<sub>3</sub> analyzer. Rates of NO production on a dry soil mass basis (μg N kg<sup>-1</sup> h<sup>-1</sup>) were calculated from the difference between NO concentration in air upstream and downstream of the soil, the air flow rate, and the dry soil mass, as previously described (Venterea & Rolston, 2000a).

The role of autotrophic nitrification in regulating NO<sub>3</sub> production and NO production was examined using acetylene (C<sub>2</sub>H<sub>2</sub>) inhibition. In parallel with the 14-day incubation experiments, separate subsamples (10–30 g) of each sample collected in June 2001 were placed in 250-mL glass jars and treated with gas-phase C<sub>2</sub>H<sub>2</sub> on the first day of incubation. Headspace concentrations of 30–40 Pa C<sub>2</sub>H<sub>2</sub> were achieved by injecting 5 mL of 1000 Pa C<sub>2</sub>H<sub>2</sub> through a butyl rubber septum into each jar. Following C<sub>2</sub>H<sub>2</sub> addition, jars were kept sealed for 24–48 h, opened for 5–10 min to allow equilibration with ambient air, and then were sealed loosely. Soils were re-treated with C<sub>2</sub>H<sub>2</sub> a second time during the incubation (after 5–10 days). Headspace concentrations of C<sub>2</sub>H<sub>2</sub> in the range of 10–100 Pa (0.01–0.1%) have been shown to inhibit autotrophic nitrifying bacteria (e.g. Nitrosomonas sp.), which mediate the oxidation of NH<sub>4</sub><sup>+</sup> to NO<sub>2</sub> (Klemedtsson et al., 1988). Rates of net nitrification and NO production were determined in the C<sub>2</sub>H<sub>2</sub>-treated soils as for non-treated soils. Prior to measuring NO production, jars were flushed with C<sub>2</sub>H<sub>2</sub>-free air for approximately 5 min at 0.06 m<sup>3</sup> h<sup>-1</sup> in order to assure that residual C<sub>2</sub>H<sub>2</sub> levels in the jars were well below concentrations of 100 Pa shown in previous studies to interfere with NO production rate measurements (Bollmann & Conrad, 1997).

The role of chemodenitrification reactions mediated by nitrous acid (HNO<sub>2</sub>) in regulating NO production was examined in samples collected in August 2000. Separate replicate subsamples (20–30 g) from each plot were extracted for 10 min with 40 mL of 1 M KCl immediately after measurement of NO production rate. Extracts were centrifuged at 3000 rpm for 20 min and adjusted to pH 7.5–8.5 using dilute sodium hydroxide to minimize complexation of NO<sub>3</sub> with soil organic matter. Extracts were analyzed for NO<sub>2</sub> with a Shimadzu UV-1601 spectrophotometer within 4 h of extraction (Vandenabeele et al., 1990). Soil pH was measured on a separate portion of the subsample used for the NO production rate and NO<sub>2</sub> measurements. Calculated HNO<sub>2</sub> concentrations were determined using the acid dissociation constant (pKa = 3.3), soil pH, and NO<sub>2</sub> – N concentrations as previously described (Venterea & Rolston, 2000a).

Within each forest type, the effect of N addition was evaluated using one-way analysis of variance (ANOVA) with the level of N addition as the main factor and with subplot measurements considered as treatment replicates, consistent with previous data analysis at the HF chronic-N study (Aber et al., 1993; Rainey et al., 1999; Magill et al., 2000). All statistical analysis was performed using Statgraphics (Manugistics, Rockville, MD).

**Results**

**Field NO and N<sub>2</sub>O fluxes**

Nitric oxide fluxes in the N-treated forest plots were consistently elevated above control plots during most of the sampling period (Fig. 1a). In the Pine site, NO fluxes in both the low-N and high-N plots were consistently elevated above the control plot. In contrast, NO fluxes in the Hardwood low-N plot were not different from the control plot, while fluxes in the high-N plot were elevated and similar in magnitude to fluxes in the Pine high-N plot (50–200 μg N m<sup>-2</sup> h<sup>-1</sup>). Using mean monthly plot values over the entire sampling period, NO flux was positively correlated (P < 0.05) with soil temperature at the 10-mm depth (r<sup>2</sup> = 0.34) and soil temperature at the 50-mm depth (r<sup>2</sup> = 0.30–0.33) within each of the N-treated Pine plots. In contrast, no significant correlation with soil temperature was evident in the N-treated Hardwood.
plots, or within either of the control plots. There were no significant correlations between soil water content and NO fluxes in either forest.

Fluxes of N$_2$O were generally $< 10 \mu$g N m$^{-2}$ h$^{-1}$ (Fig. 1b). Although there were significant differences in N$_2$O fluxes between treatment and control plots at certain times, particularly in the Pine forest, these differences were not as dramatic or consistent as for NO flux. In general, less seasonal variation in N$_2$O fluxes was evident as compared to the NO flux data. In plots displaying elevated NO fluxes (Fig. 1a), the ratio of N$_2$O to NO flux (N$_2$O:NO ratio) was generally $< 0.10$. The mean N$_2$O:NO ratio ranged from 0.04 in the Hardwood high-N plot to 0.06 and 0.08 in the Pine low-N and high-N plots, respectively, and was not correlated with organic or mineral soil water content in any of the plots or overall. In the control plots, N$_2$O fluxes were generally less than zero. No significant correlations between N$_2$O flux and soil temperature or mineral soil water content were found. Organic soil water content and N$_2$O flux tended to be negatively correlated. The negative relationship was significant ($P < 0.05$) in the Hardwood control plot ($r^2 = 0.56$) and the Pine low-N plot ($r^2 = 0.41$).

Nitric oxide fluxes measured 24h after the NH$_4$NO$_3$ application of 29 May 2001 were approximately five and seven times higher than pre-application fluxes in the Pine low-N and high-N plots, respectively (Fig. 2a). An increase of approximately 25% above pre-application levels was observed in the Hardwood high-N forest after 24h, although this change was within the range of day-to-day variation observed prior to application and 24–48h after application. Fluxes returned to pre-application levels at both sites 48h following N addition.

Soil NO fluxes measured 6 and 30h following a simulated 25-mm rainfall event in mid-August 2001, and 24h following a similar event in late-October 2001 displayed no detectable differences from pre-wetting fluxes (Fig. 2b). Soil temperatures ($T_s$) within each plot varied by 2–6 °C at the 10-mm depth, and by only 1–2 °C at the 50-mm depth, during daylight hours on 3 July and 27 October 2001 (Fig. 3). In the July experiment, NO fluxes within each plot were significantly ($P < 0.05$) and positively correlated with $T_s$ at both the 10-mm depth

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**Fig. 1** Fluxes of (a) NO and (b) N$_2$O during June 2000 through November 2001. Values are the means of three measurements in the control, low-N, and high-N treatment plots at approximately monthly intervals. Symbols indicate if the low-N (#) or high-N (*) plot values are significantly different from the control at any time based on ANOVA with least significant differences multiple range test. # or *$P < 0.05$; ## or **$P < 0.01$; ### or ***$P < 0.001$.

**Fig. 2** Response of NO flux to (a) NH$_4$NO$_3$ applied on 29 May 2001 and (b) simulated rainfall events on 18 August and 27 October 2001.
(r² = 0.35–0.60) and 50-mm depth (r² = 0.46–0.91) except in the Hardwood high-N plot (r² < 0.1, P > 0.5). In the October experiment, only the Hardwood low-N plot displayed a positive correlation with Tₘ (r² = 0.69–0.73, P < 0.05), while the Pine low-N plot displayed a significant negative correlation with Tₘ at the 10-mm depth (r² = 0.54, P < 0.05). In most cases, NO fluxes measured toward the middle of the day were intermediate in magnitude between early morning and afternoon fluxes (Fig. 3).

**Soil analysis**

Net nitrification rates (NRs) determined in 14-day laboratory incubations of soils collected in August 2000 and June 2001 displayed the same pattern demonstrated by field NO fluxes over the course of the study: i.e., NRs in the Hardwood and Pine high-N plots, and Pine low-N plots, were elevated above control plots (Fig. 4a). Nitric oxide production rates in organic soils displayed the same pattern as field NO fluxes and NRs, and a similar though not entirely consistent pattern was evident in NO production rates in mineral soils (Fig. 4b). The application of acetylene in parallel soil incubations resulted in near-complete inhibition of both nitrification and NO production in mineral and organic soil (Fig. 4a, b). Soil NO₃⁻ concentrations displayed the same pattern as laboratory-measured NRs, while soil NH₄⁺ concentrations and N mineralization rates displayed no consistent pattern (Table 1).

Nitrile (NO₃⁻) levels could not be detected (<2 µg N kg⁻¹) in any of the organic soils, but were detectable (3–11 µg N kg⁻¹) in mineral soils collected August 2000. Nitric oxide production rate (P_NO) in Pine mineral soils was positively correlated (r² = 0.66, P < 0.001) with NO₃⁻ concentration (r² < 0.05 in Hardwood soils). A kinetic model following that developed by Venterea & Rolston (2000a, b) and represented by

\[ P_{NO} = a[HNO_2]^b \]  

was fitted to the data by non-linear regression, where a is the NO production rate coefficient for HNO₂-mediated NO production, and b is the empirical reaction order. The obtained regression models were statistically significant (P < 0.001) with reaction order values of 3.6 and 2.4 for Hardwoods (r² = 0.84) and Pines (r² = 0.90), respectively (Fig. 5). Soil pH in mineral and organic soils was lower in the N-treated plots compared to the control plots, and were positively correlated with the annual rate of N deposition (Table 2).
Discussion

Patterns in N oxide gas emissions

Our findings extend previous results from the Harvard Forest chronic-N addition study to include elevated NO flux along with enhanced NO$_3^-$ production and mobility as a response to persistent N inputs (Bowden et al., 1991; Magill et al., 2000). Soil NO fluxes were of similar magnitude to the other major N loss mechanism at HF, i.e., NO$_3^-$ leaching below the root zone. Typical summertime NO fluxes (100–200 $\mu$g N m$^{-2}$ h$^{-1}$, Fig. 1a) were equivalent to 15–45% of annual NO$_3^-$ leaching rates in the high-N plots and >100% of NO$_3^-$ losses in the Pine low-N plots estimated for 1996 (Magill et al., 2000). Estimates of NO$_3^-$ losses for the period 1996–2001, which are currently being prepared for publication (A. Magill, personal communication), will allow for contemporaneous comparisons of NO$_3^-$ and NO losses.

The differential response in the two forest types observed here is consistent with trends previously observed at HF. As of 1996, significant increases in nitrification rates were observed in the Pine low-N plot and in both the Pine and Hardwood high-N plots, while NO$_3^-$ leaching was observed only in the high-N plots (Magill et al., 2000). The current findings provide new indication that the Pine low-N plot has advanced beyond initial stages of N saturation, as it is now exhibiting responses similar to both high-N plots with respect to enhanced nitrification and NO emissions (Aber et al., 1998). The Hardwood low-N plot continues to resist changes in N cycling rates.

The elevated NO fluxes observed at HF are similar to fluxes of 20–130 $\mu$g N m$^{-2}$ h$^{-1}$ in spruce-dominated plots in Germany which receive >30 kg N ha$^{-1}$ yr$^{-1}$ of deposition (Butterbach-Bahl et al., 1997). Emissions of NO from beech-dominated plots in this forest were lower (6–47 $\mu$g N m$^{-2}$ h$^{-1}$). Thus, the pattern of higher NO
Table 1  Concentrations of nitrate-N and ammonium-N and net N mineralization rates in soils from Harvard Forest chronic-N plots

<table>
<thead>
<tr>
<th>Month</th>
<th>Nitrate-N concentration (mg N kg⁻¹)</th>
<th>Ammonium-N concentration (µg N kg⁻¹)</th>
<th>Net N mineralization rate (µg N kg⁻¹ h⁻¹)</th>
</tr>
</thead>
<tbody>
<tr>
<td>June 2000</td>
<td>Control Low-N</td>
<td>0.22 (0.08)</td>
<td>0.30 (21)</td>
</tr>
<tr>
<td></td>
<td>Low-N</td>
<td>0.24 (0.09)</td>
<td>0.31 (0.06) n.a.</td>
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<tr>
<td></td>
<td>High-N</td>
<td>0.28 (0.10)</td>
<td>0.33 (0.07) n.a.</td>
</tr>
<tr>
<td>June 2001</td>
<td>Control Low-N</td>
<td>0.46 (0.13)</td>
<td>0.54 (0.20) n.a.</td>
</tr>
<tr>
<td></td>
<td>Low-N</td>
<td>0.58 (0.16)</td>
<td>0.63 (0.23) n.a.</td>
</tr>
<tr>
<td></td>
<td>High-N</td>
<td>0.66 (0.18)</td>
<td>0.70 (0.24) n.a.</td>
</tr>
</tbody>
</table>

Fig. 5  Laboratory-measured rates of NO production in mineral soils sampled August 2000 vs. calculated nitrous acid (HONO) concentrations. Lines are results of non-linear regression analysis using kinetic models in the form of Eqn 2, P < 0.001.

fluxes in the N-impacted coniferous forest as compared to the deciduous forest in Germany is similar to the pattern observed in the low-N plots at HF, and is consistent with the idea that deciduous forests in general may have a greater capacity to retain N inputs than coniferous forests with similar land-use histories (Aber et al., 1998).

Questions remain as to the factors controlling differences in response between the two forests. Data through 1996 show that vegetation in the Hardwood plots had assimilated only slightly more of the N added since 1988 than vegetation in the Pine plots. Based on ecosystem N budget calculations through 1996, the major N sink is believed to be some component of soil organic matter, accounting for ≥70% of inputs (Magill et al., 2000). Consideration of our first-time estimates of annual NO emissions would reduce the proportion of inputs attributed to soil retention by modest amounts (~3–8%, see below). Our data do not suggest that net N mineralization rates (MRs) are limiting nitrification rates in the Hardwood low-N plot by substrate limitation, as there were no consistent trends or differences in soil NH₄ or MRs (Table 1). This is consistent with previous findings, which have shown no response, or even a negative response, in MRs to N additions (Aber et al., 1998). Thus, possible explanations for the delayed response in nitrification in the Hardwood forest compared to the Pine forest include those which have been previously discussed in detail by Aber et al. (1998), i.e., higher rates of N assimilation by heterotrophic microbes, abiotic processes, and/or mycorrhizae, all of which may compete
with nitrifying microbes for N, and/or differences in land-use history. Elucidation of these mechanisms remains a critical goal of ongoing research (e.g., Bernston & Aber, 2000).

It is certainly not clear to what extent the responses observed after 12–13 years of artificially high rates of N addition in the high-N plots mimic the cumulative effects which might be observed over longer periods with deposition rates more representative of impacted areas. However, data from the high-N plots at HF have been valuable in suggesting that the two forest types have finite, and differential, capacities to retain added N. In the absence of the high-N treatments, there would be no basis for estimating the limits of N retention in the Hardwood forest (Aber et al., 1998). The results also support general hypotheses about the process of N saturation, whereby N oxide emissions and NO$_3^-$ leaching are proposed as ecosystem responses which are likely to be non-linear with respect to cumulative N inputs (Aber et al., 1998; Fenn et al., 1998), as demonstrated here.

Monthly flux measurements may not be sufficient to very accurately estimate contributions to the overall ecosystem N mass balance because of generally high temporal variability of N oxide fluxes (e.g., Williams et al., 1992). Nonetheless, our short-term data (Figs 2, 3) do not indicate that variability at time-scales of days or hours was so extreme as to completely preclude such estimation in the absence of additional data. We therefore applied a set of assumptions used in a previous study to estimate annual N$_2$O emissions at HF (Bowden et al., 1991), i.e., each sampling date was treated as the midpoint of a sampling period during which gas fluxes were assumed to equal the mean plot flux measured on that date. The estimated total NO emissions were similar in the Hardwood and Pine high-N plots, representing 3.0% (equivalent to 4.7 kg N ha$^{-1}$ yr$^{-1}$) and 3.7% (5.9 kg N ha$^{-1}$ yr$^{-1}$), respectively, of total N inputs (background plus experimental) during June 2000–November 2001. The integrated NO mass flux from the Pine low-N plot represented the greatest proportion (8.3% or 4.8 kg N ha$^{-1}$ yr$^{-1}$) of total N inputs. Emissions from the Hardwood and Pine control plots were 1.3 and 2.3%, respectively, of total inputs, while NO emitted from the Hardwood low-N plot represented the smallest proportion of inputs (0.3% or 0.16 kg N ha$^{-1}$ yr$^{-1}$). These amounts compare to NO$_3^-$ leaching losses representing approximately 2% of inputs to the low-N plots in both forests, and 15 and 4% of inputs to the high-N plots in the Pine and Hardwood forests, respectively, during 1988–1996 (Magill et al., 2000). Total N$_2$O emissions were <5% of total NO emissions and <0.3% of total N inputs in all plots.

Significant positive correlations ($P < 0.01$) were found between net nitrification rates and total N oxide gas fluxes (i.e. NO + N$_2$O) measured at approximately the same time as soil sample collection ($r^2 = 0.49$ in mineral and 0.65 in organic soil). Similarly, total N oxide gas flux was correlated ($P < 0.01$) with soil NO$_3^-$ concentrations in mineral ($r^2 = 0.58$) and organic soil ($r^2 = 0.52$). These relationships are consistent with the hole-in-the-pipe (HIP) model which describes N oxide gas flux primarily as a function of N cycling rates and other indices of N availability, with partitioning between NO and N$_2$O based on water-filled pore space (Davidson et al., 2000). Soil NH$_4^+$ concentrations and net N mineralization rates were not correlated ($P > 0.42$) with total N oxide flux. These findings are also consistent with the predominance of nitrification, and not N mineralization, as the main process regulating N oxide emissions at HF.

The short-term responses to N addition observed in the Pine stand following N application (Fig. 2a) represent experimental artefacts of the N treatments, since actual N deposition would occur in more dilute inputs distributed over time. These results point out potential limitations in N deposition simulation studies, some of which have used N application methods similar to those at HF (Kahl et al., 1993; Peterjohn et al., 1998), while others have used more frequent, dilute applications (Klemetsson et al., 1997). No other studies, to our knowledge, have examined such short-term responses in NO flux. We
hypothesize that the pulse of NO flux in the Pine plots was due to transient increases in the production of NO\(_2\) by NH\(_4\)-oxidizing nitrifiers accompanied by a lag in the response of NO\(_2\)-oxidizing nitrifiers (Venterea & Rolston, 2000).

**Sources of NO production**

Simultaneous inhibition of nitrification and NO production by acetylene (Fig. 4) indicated that NO production was due to reactions accompanying the autotrophic oxidation of NH\(_4\) to NO\(_2\) (Davidson et al., 1986; Klemmedtsson et al., 1988). Similar conclusions regarding nitrification as the source of NO production in tropical forest soils have been obtained using similar techniques (Davidson, 1992a,b). The correlation between NO production and HNO\(_2\) (Fig. 5) further suggests abiotic decomposition of HNO\(_2\) (i.e. ‘chemodenitrification’) during nitrification as an important NO-production mechanism. Nitrous acid-mediated reactions have been indicated as an important mechanism of NO production in agricultural soils (Blackmer & Cerrato, 1986; Venterea & Rolston, 2000a,b) as well as forest and grassland soils (Davidson, 1992a,b; Yamulki et al., 1997). Since chemodenitrification requires the protonation of nitrification-derived NO\(_2\), these results together with the pH data (Table 2) suggest that acidity may be an important factor in promoting N losses in forests impacted by atmospheric deposition. Thus, management practices such as amending soils with lime, which has been shown to reduce emissions of N\(_2\)O from forest soils (Brumme & Beese, 1992; Klemmedtsson et al., 1997), may also be effective in reducing NO emissions. Liming of spruce plots in Southern Germany did in fact result in a 25–30% reduction in NO fluxes (Butterbach-Bahl et al., 1997).

Previous studies have also considered the role of soil organic matter (SOM) in enhancing HNO\(_2\)-mediated NO production (Stevenson et al., 1970). Venterea & Rolston (2000a) found a positive correlation between the NO production rate coefficient (\(a\) in Eqn [2]), and organic C in three agricultural soils containing 0.3–1.4% organic C. Organic C concentrations in mineral soils at HF are 5–8% (Aber et al., 1993), and the rate coefficients observed in HF mineral soils are approximately 6–8 times greater than the corresponding \(a\) values in these agricultural soils (Venterea & Rolston, 2000a). Thus, the current data are consistent with a mechanism of NO production involving reactions between HNO\(_2\) and SOM.

There are methodological challenges in further elucidating the mechanisms of NO production in forest soils. Nitrite levels were not detectable in the organic soils, although organic soils generally displayed higher rates of NO production than mineral soils (Fig. 4b). Organic matter, in particular compounds containing phenolic groups, can interfere with NO\(_2\) analysis in soil extracts under acidic conditions (Vandenabeele et al., 1990). While we attempted to minimize these interferences, it is not known to what extent these effects influenced our determination of NO\(_2\) levels. Thus, the importance of chemodenitrification in the organic soils cannot be eliminated. Conversely, the relationships in Fig. 5 do not preclude the importance of other sources of NO production coupled to nitrification, such as biological NO\(_2\) reduction (Conrad, 1995). The statistical relationship between HNO\(_2\) concentrations and NO production in the Hardwood mineral soils was largely influenced by a single data point exhibiting the highest rate of NO production and HNO\(_2\) concentration (Fig. 5). Thus, the Hardwood kinetic data are not as convincing as the Pine data with respect to the role of HNO\(_2\) in mediating nitrification-derived NO production. This could possibly have been related to variation in soil organic matter quantity or quality within mineral soil samples from the Hardwood plots, reactions between NO\(_2\) and SOM which may act to fix some fraction of the NO\(_2\)/HNO\(_2\) (Smith & Chalk, 1980), and/or interferences in the measurement of low NO\(_2\) levels in organic-rich soils, as discussed above.

**Climatic factors**

Soil temperature accounted for 30–34% of the overall variance in monthly NO flux within each Pine N-treated plot, but no significant correlations were observed in the Hardwood monthly data. While some previous studies have shown a positive correlation between NO flux and soil temperature (e.g. Williams et al., 1992), recent experimental and modeling studies have suggested that NO consumption, as well as production, exert significant control over net NO emissions, and also that the sensitivity of NO consumption rates to soil temperature may significantly impact overall temperature effects on net NO emissions (Stark et al., 2002; Venterea & Rolston, 2002). Thus, our data may indicate a greater importance of NO consumption in the Hardwood plots with respect to temperature controls. Increased NO consumption with temperature may also explain the decreases in NO flux observed during the afternoon of 27 October 2001 when soil temperatures at the 10-mm depth continued to increase (Fig. 3).

There were no significant correlations between soil water content and NO fluxes. This is consistent with the fact that soil water content may have both negative and positive effects on NO emissions, i.e., increased water availability may enhance nitrification (Schmidt, 1982) while at the same time impeding gaseous diffusion and thereby promoting NO transformation within the soil.
matrix prior to release at the soil surface (Davidson et al., 2000; Venterea & Rolston, 2000c). While rain-induced pulses in NO emissions persisting for several days have been observed in seasonally dry ecosystems, including tropical forests and savannas, and chaparral and grasslands in Mediterranean climates (e.g. Davidson 1992a; Otter et al., 1999), similar effects in temperate forests have yet to be demonstrated, and were not supported by our data (Fig. 2b). The amount of water added in the present experiments (25 mm) was selected based on the frequency and quantity of rainfall events occurring at HF. During 2000–2001, nine rainfall events exceeded 25 mm in a 24-h period, but all were <37 mm, while one event produced 53 mm in a 24-h period. Thus, our results suggest that rain-induced pulses of gas emissions were not important in controlling overall NO emissions during the study period.

Acknowledgements

The authors gratefully acknowledge Sabrina LaFave and Jessica Kriebel for assistance with the laboratory analyses, and Rosalie Cabral, Mark Venterea, and Michael Hannigan for assistance with various field activities. We also thank Eric A. Davidson and two anonymous reviewers for valuable comments, which helped to significantly improve earlier versions of the manuscript. This work was funded by EPA NCERQA (Grant R827674).

References


